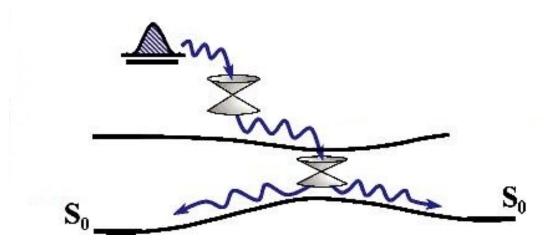
Theoretical Photochemistry WiSe 2016/17

Lecture 10



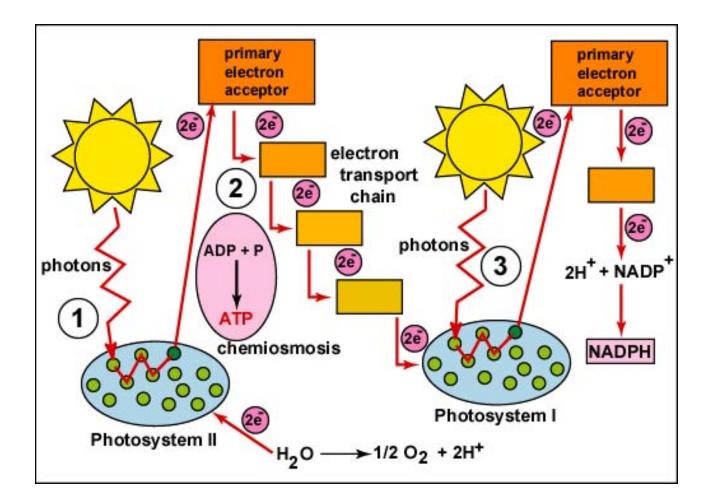
Irene Burghardt (burghardt@chemie.uni-frankfurt.de)

 $http://www.theochem.uni-frankfurt.de/teaching/ \longrightarrow Theoretical Photochemistry$

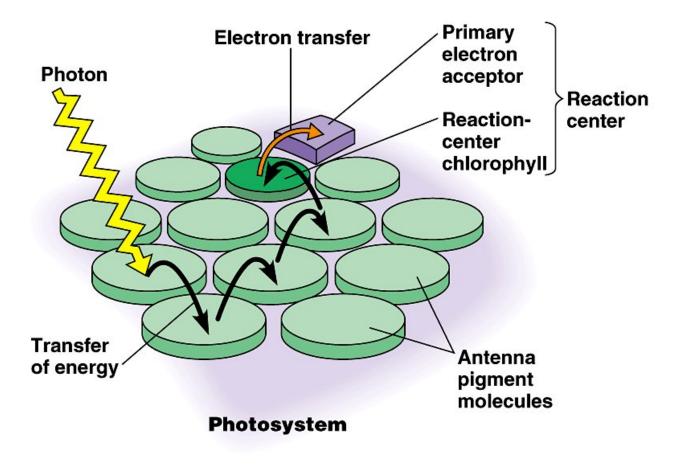
Topics

- **1. Photophysical Processes**
- 3. Wavepackets
- 5. The Franck-Condon picture of electronic transitions
- 6. What do we measure experimentally?
- 2. The Born-Oppenheimer approximation
- 4. Beyond Born-Oppenheimer non-adiabatic transitions
- 7. Conical intersections
- 8. Examples: Ethene, Protonated Schiff Bases (Retinal), Azobenzene
- 9. Intermolecular energy transfer, light-harvesting, and Förster theory
- 10. Dynamics: trajectories or wavefunctions?
- 11. Coherence and Dephasing
- 12. Non-linear optical spectroscopy: calculation of spectroscopic signals
- 13. Solvent/environmental effects

Photoinduced intermolecular energy transfer



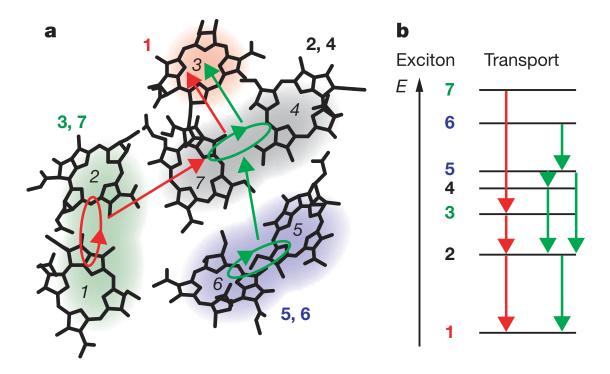
Light-harvesting systems (LH's)



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Exciton transfer in the FMO complex

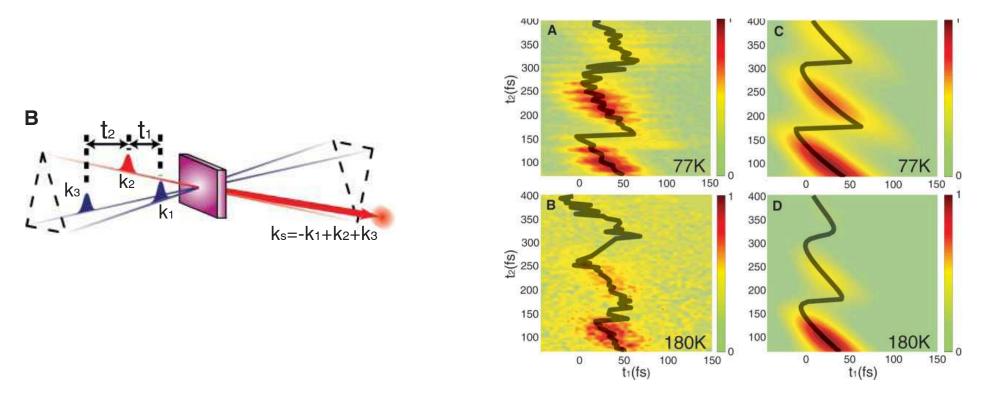
Brixner & collaborators, Nature 434, 625 (2005)



- FMO = Fenna-Matthews-Olson bacteriochlorophyll a (BChl) protein of green sulphur bacteria
- antenna system that collects light and channels excitation to a reaction center where charge transfer takes place

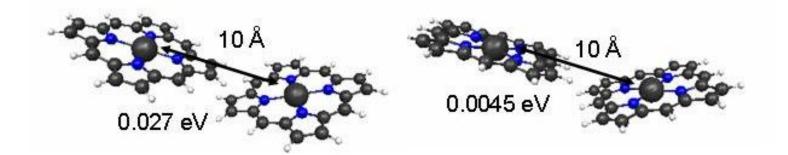
"Coherence dynamics in photosynthesis: protein protection of excitonic coherence"

Lee, Cheng, Fleming, Science 316, 1462 (2007)



• 2CECPE = two-color electronic coherence photon echo experiment

Multi-chromophoric systems



• Two basic (single-excited) configurations:

 $|\phi_1
angle = |e^{(1)}
angle \otimes |g^{(2)}
angle \qquad |\phi_2
angle = |g^{(1)}
angle \otimes |e^{(2)}
angle$

- For two identical, non-interacting monomers, these configurations are degenerate
- once an excitonic coupling is included, the degeneracy is removed
- Frenkel exciton state = superposition of these configurations:

 $|\Psi_{
m exciton}(t)
angle = c_1(t)|\phi_1
angle + c_2(t)|\phi_2
angle$

What is the excitonic coupling?

 Coulomb coupling matrix element between donor (D) and acceptor (A) permitting transitions between single-excited configurations:

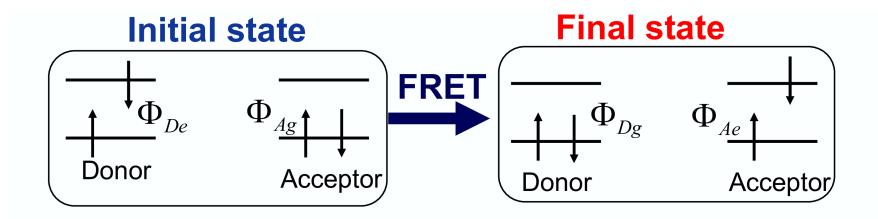
 $j = \langle e^D g^A | V_{
m Coulomb} | g^D e^A
angle$

• this can be re-written in terms of coupled transition densities $\rho_A^{(ge)}$ and $\rho_D^{(eg)}$ (noting that D/A exchange interactions have been disregarded!):

$$j = \int d{
m r}_D^3 d{
m r}_A^3 rac{
ho_D^{(eg)}({
m r}_D)
ho_A^{(ge)}({
m r}_A)}{|{
m r}_D - {
m r}_A|}$$

• at large D-A distances, one can approximate transition densities by transition dipoles: $\rho_D^{(eg)} \longrightarrow \mu_D^{(eg)}$

Dipole Approximation

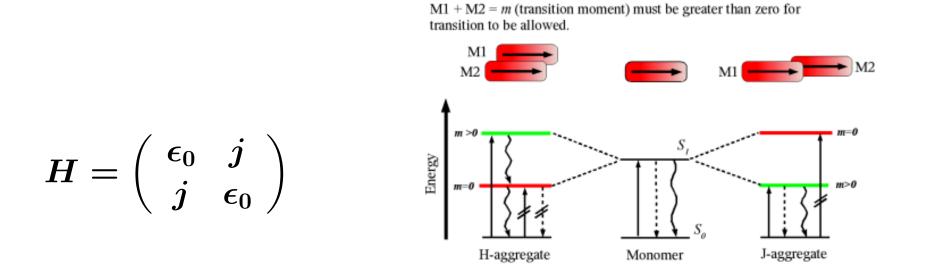


$$\begin{split} \langle \psi_D^* \psi_A^0 | V_{DA}^{\text{dip}} | \psi_D^0 \psi_A^* \rangle &= \langle \psi_D^* \psi_A^0 | \frac{\hat{\mu}_D \cdot \hat{\mu}_A}{|r_{DA}|^3} - 3 \frac{(r_{DA} \cdot \hat{\mu}_D)(r_{DA} \cdot \hat{\mu}_A)}{|r_{DA}|^5} | \psi_D^0 \psi_A^* \rangle \\ &= \kappa_{DA} \frac{\langle \psi_D^* | \hat{\mu}_D | \psi_D^0 \rangle \langle \psi_A^0 | \hat{\mu}_A | \psi_A^* \rangle}{|r_{DA}|^3} \end{split}$$

 κ_{DA} = orientational factor

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J and H Aggregates



- j < 0: J-aggregates, j > 0: H-aggregates
- absorption spectra are red-shifted (J-aggregates) and blue-shifted (H-aggregates)

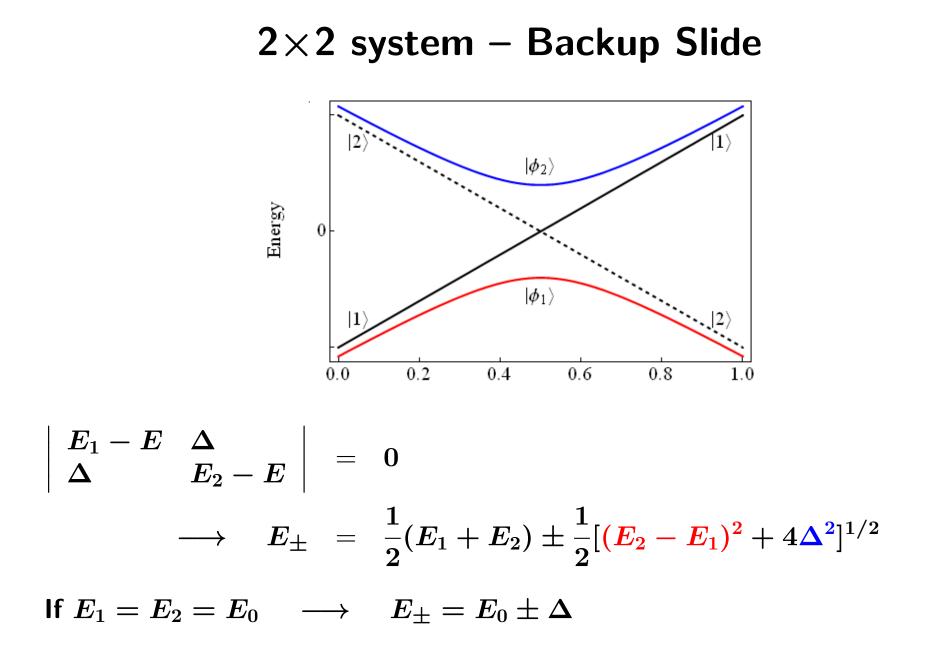
J and H Aggregates

Solve for the eigenvalues and eigenvectors of the matrix Hamiltonian (in the basis of excitonic configurations $|\phi_1\rangle, |\phi_2\rangle$):

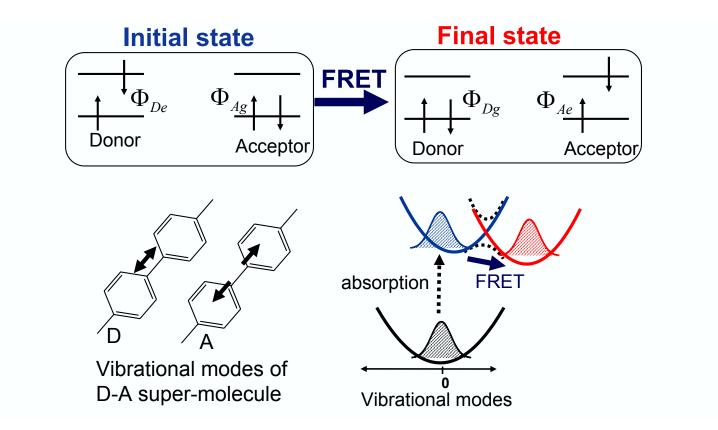
$$H=\left(egin{array}{cc} \epsilon_0 & j \ j & \epsilon_0 \end{array}
ight)$$

eigenvalues: $E_{\pm} = \epsilon_0 \pm j$ eigenvectors: $|\phi_{\pm}\rangle = 1/\sqrt{2}(|\phi_1\rangle \pm |\phi_2\rangle)$

- for the eigenvector $|\phi_+\rangle$, the monomer transition dipole moments $M_i = \langle e^{(i)} | \hat{\mu} | g^{(i)} \rangle$, i = 1, 2, add up constructively, while the opposite is the case for the $|\phi_-\rangle$ eigenvector \longrightarrow "bright" vs. "dark" states
- the sign of the coupling j determines which eigenvalue is the lower one. Therefore, the sign of the coupling j also determines whether the upper or lower state is dark – distinction between J and H aggregates!



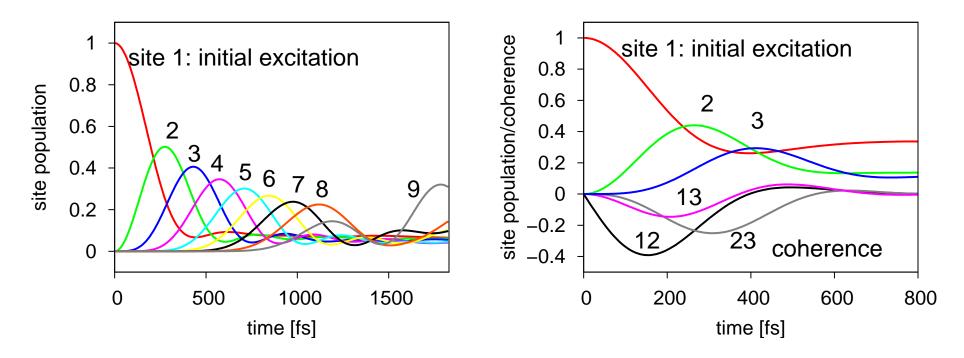
Vibronic coupling picture of excitation energy transfer (EET)



• FRET = fluorescence resonance energy transfer (equivalent to EET)

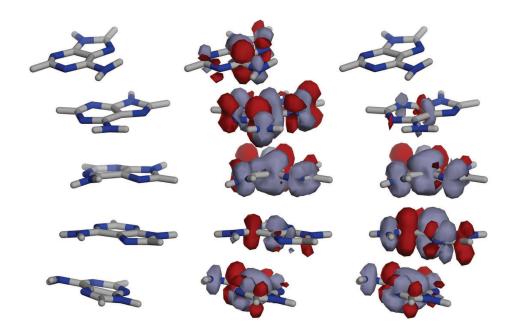
Exciton transfer can be coherent and ultrafast

quantum dynamics simulation:



- site-site excitation energy transfer $|\phi_n
 angle \longrightarrow |\phi_{n+1}
 angle$
- transfer is mediated by coherences $|\phi_n\rangle\langle\phi_{n+1}|$
- but the environment could rapidly induce "decoherence"
- coherent transfer can be observed experimentally!

Excitonic eigenstate picture



electronic densities of stacked adenine pentamer corresponding to two $\pi\pi^*$ states with different degrees of localisation

Bittner & co-workers, in: *Energy Transfer Dynamics in Biomaterial Systems*, Burghardt et al. (eds), Springer (2009).

• the excitonic eigenstates are obtained by diagonalizing the exciton Hamiltonian, as a function of the nuclear coordinates:

$$|\Psi_{
m exciton}(R;t)
angle = \sum_n c_n(R,t) |\phi_n
angle$$

Can we use perturbative limits?

consider site-site coupling (J) vs. electron-phonon coupling (κ)

• case 1: $J \ll \kappa$

Förster theory – non-coherent hopping between sites

e.g., Scholes & collaborators, J. Phys. Chem. B, 113, 656 (2009): EET in semiconducting polymers

• case 2: $\kappa \ll J$

excitonic eigenstates in a vibrational bath

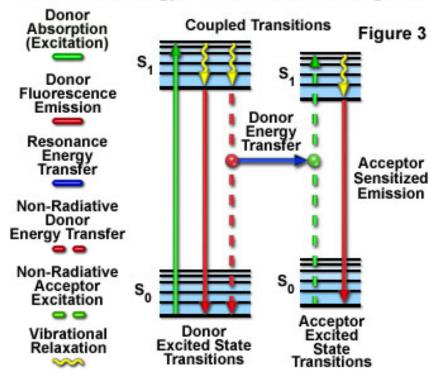
e.g., Abramavicius & Mukamel, J. Chem. Phys. 133, 064510 (2010): EET in Photosystem II

• but in most systems of interest: $\kappa \sim J$

in principle, we need the full dynamics on the Born-Oppenheimer surfaces of the oligomer/aggregate species

Excitation Energy Transfer (EET): Förster rates

Resonance Energy Transfer Jablonski Diagram



- resonant donor-acceptor transfer
- also denoted FRET = fluorescence resonance energy transfer
- interacting transition dipole moments
- standard description: Förster rate

Scholes & collaborators, J. Phys. Chem. B, 113, 656 (2009)

Fluorescent markers in biological applications

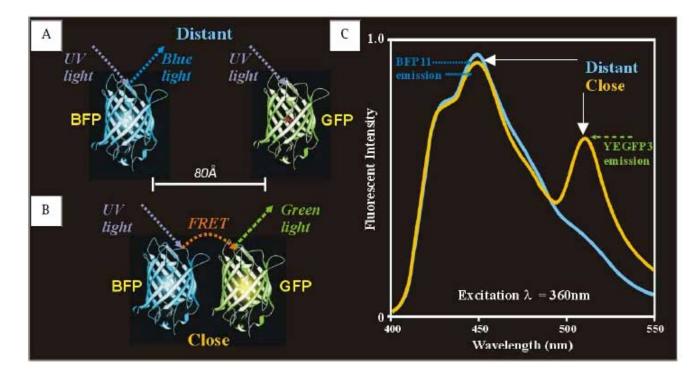
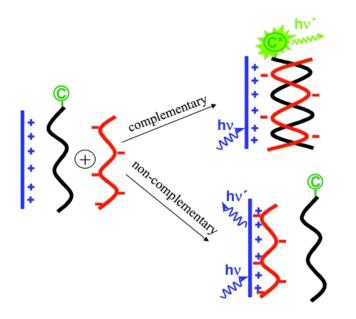


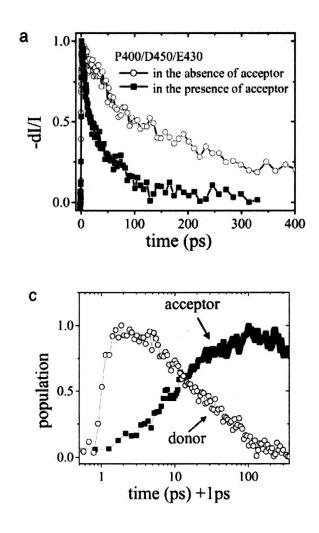
Figure 1. The use of GFP variants to produce FRET. In A, Distant; UV light excites BFP (donor fluorophore) to emit blue light (peak emission = 450nm), but GFP (acceptor) is not close enough to draw energy from the excited donor. In B, Close; proximity of GFP to BFP allows non-radiative energy transfer, the stimulated BFP exciting GFP to fluoresce green (peak emission = 510nm). Figure 1C shows the expected emission spectra when the GFP fluorophores are either close or distant.

Time-resolved FRET

Heeger & co, PNAS 101, 11634 (2004), PNAS 102, 530 (2005)



CCP (cationic conjugated polymer) → PNA-C (peptide nucleic acid + chromophore C=fluorescein)



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Derivation of Förster theory, in a nutshell: Fermi's Golden Rule + dipole-dipole interaction

thermally averaged rate (2nd order perturbation theory):

$$k_{DA} = \frac{2\pi}{\hbar} \sum_{\text{D/A vib. states}} f(E_D^*) f(E_A^0) \left| \langle \psi_D^* \psi_A^0 | V_{DA}^{\text{dip}} | \psi_D^0 \psi_A^* \rangle \right|^2 \delta(E_D^* + E_A^0 - E_A^* - E_D^0)$$

$$\begin{split} \langle \psi_D^* \psi_A^0 | V_{DA}^{\text{dip}} | \psi_D^0 \psi_A^* \rangle &= \langle \psi_D^* \psi_A^0 | \frac{\hat{\mu}_D \cdot \hat{\mu}_A}{|r_{DA}|^3} - 3 \frac{(r_{DA} \cdot \hat{\mu}_D)(r_{DA} \cdot \hat{\mu}_A)}{|r_{DA}|^5} | \psi_D^0 \psi_A^* \rangle \\ &= \kappa_{DA} \frac{\langle \psi_D^* | \hat{\mu}_D | \psi_D^0 \rangle \langle \psi_A^0 | \hat{\mu}_A | \psi_A^* \rangle}{|r_{DA}|^3} \end{split}$$

Now identify the ingredients of this formula with the expressions for the donor emission / acceptor absorption spectra \rightarrow spectral overlap ²⁰

Fermi's Golden Rule

Transition probability between quantum states that are subject to a perturbation

$$\hat{H}(t) = \hat{H}_0 + \hat{V}(t)$$

e.g., $\hat{V}(t) = -\hat{\mu} E_0 \left(e^{i\omega t} + e^{-i\omega t} \right)$ "perturbation"

Transition rate between two states $a \rightarrow b$ (in 2nd order perturbation theory):

$$\Gamma_{a
ightarrow b}=rac{2\pi}{\hbar}|\langle\psi_b^{(0)}|\hat{\mu}|\psi_a^{(0)}
angle|^2\;\delta(E_b^0-E_a^0\pm\hbar\omega)$$

where $E_b^0 - E_a^0 = \hbar \omega_{ba}$ - resonance condition!

Förster theory = rate theory for resonant energy transfer between two dipoles

$$k_{DA}(r_{DA}) = k_D^{\mathrm{rad}} \Big(rac{r_F}{r_{DA}}\Big)^6$$

$$r_F^6 = rac{9c^4 \,\kappa_{DA}^2}{8\pi\eta^4} \int_0^\infty d\omega \; rac{1}{\omega^4} I_D(\omega) lpha_A(\omega)$$

"Förster radius" $r_F \propto$ orientational factor imes spectral overlap

 $I_D(\omega) =$ donor emission spectrum

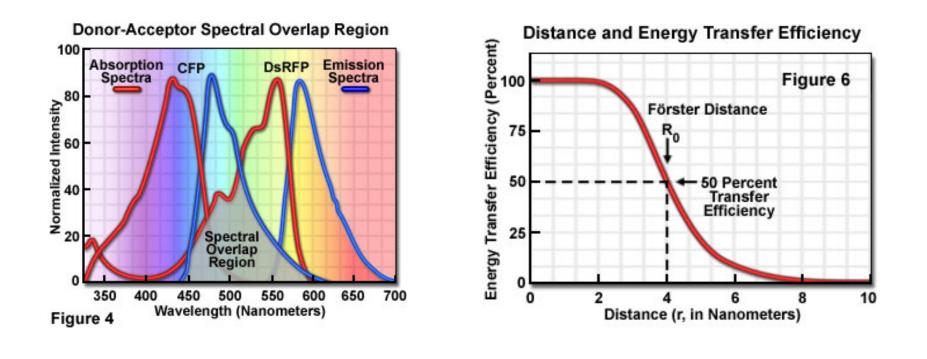
 $\alpha_A(\omega) =$ acceptor absorption spectrum

 κ_{DA} = orientational factor:

$$\kappa_{DA} = \sin\theta_D \sin\theta_A \cos\Phi_{ARD} - 2\cos\theta_D \cos\theta_A$$

Standard Förster expression: isotropic average $\kappa_{DA}^2 = 2/3$

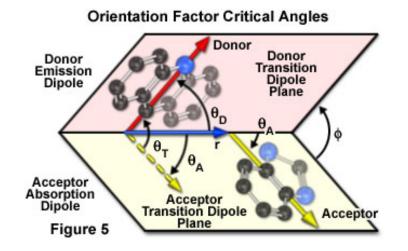
Förster theory, cont'd



$$k_{DA}(r_{DA})=k_D^{
m rad}\Big(rac{r_F}{r_{DA}}\Big)^6 ~~;~~ r_F^6=rac{9c^4\,\kappa_{DA}^2}{8\pi\eta^4}\int_0^\infty d\omega~rac{1}{\omega^4}\,I_D(\omega)lpha_A(\omega)$$

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Isotropic orientational average often inappropriate

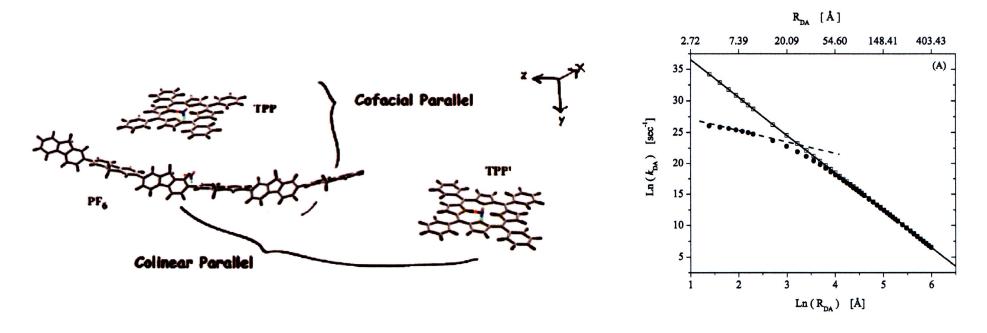


- standard Förster expression works if rotational motion is very fast as compared with transfer rate
- in general: compute $k_{DA}(r_{DA}, \theta_{DA})$
- if transfer is much faster than rotational motion: static average

Non-Förster behavior due to orientation effects

Wong, Bagchi, and Rossky, J. Phys. Chem. A, 108, 5752 (2004)

polyfluorene (PF_6) \rightarrow tetraphenylporphyrin (TPP)



The Förster rate is recovered at large distances, $\sim 100~\AA$ At shorter distances, r^{-2} gives a better fit

We'd expect deviations from Förster theory, for various reasons:

• if EET time scales are short, and similar to the time scale of molecular rearrangements: exciton dynamics picture necessary

- at short distances, exchange effects need to be accounted for (Dexter, 1953) \rightarrow exponential variation of the rate with r
- for extended donor/acceptor entities, the point dipole approximation is often not sufficient → multipole expansions
- for molecules near surfaces, the energy transfer is carried by (i) plasmons (collective electronic excitation modes) and (ii) electron-hole excitations