

Lösungen des Übungsblattes 9 zur Vorlesung Theoretische Chemie I WS 2018/19 – Übungsblatt 9

1. Die Schrödingergleichung des Wasserstoffatoms

$$\left(-\frac{\hbar^2}{2m_e} \frac{\partial^2}{\partial x_e^2} - \frac{\hbar^2}{2m_N} \frac{\partial^2}{\partial x_N^2} - \frac{e^2}{4\pi\epsilon_0 |x_e - x_N|} \right) \Psi(x_e, x_N) = E \Psi(x_e, x_N) \quad (1)$$

$$X = \frac{m_e}{M} x_e + \frac{m_N}{M} x_N \quad (2)$$

$$r = x_e - x_N \quad (3)$$

a) (Referenz: Abschnitt 3.2 „Reduced mass“ in Further Information, Kapitel 3, *Molecular Quantum Mechanics*, 5th ed., Peter Atkins und Ronald Friedman)

Verwenden der Kettenregel

$$\begin{aligned} \frac{\partial \psi}{\partial x_e} &= \left(\frac{\partial \psi}{\partial X} \right) \left(\frac{\partial X}{\partial x_e} \right) + \left(\frac{\partial \psi}{\partial r} \right) \left(\frac{\partial r}{\partial x_e} \right) = \frac{\partial \psi}{\partial X} \cdot \frac{m_e}{M} + \frac{\partial \psi}{\partial r} \cdot 1 \\ &= \frac{m_e}{M} \frac{\partial \psi}{\partial X} + \frac{\partial \psi}{\partial r} \end{aligned}$$

$$\begin{aligned} \frac{\partial^2 \psi}{\partial x_e^2} &= \frac{\partial}{\partial x_e} \frac{\partial \psi}{\partial x_e} = \frac{m_e}{M} \frac{\partial}{\partial X} \left(\frac{m_e}{M} \cdot \frac{\partial \psi}{\partial X} + \frac{\partial \psi}{\partial r} \right) + \frac{\partial}{\partial r} \left(\frac{m_e}{M} \cdot \frac{\partial \psi}{\partial X} + \frac{\partial \psi}{\partial r} \right) \\ &= \frac{m_e^2}{M^2} \frac{\partial^2 \psi}{\partial X^2} + \frac{m_e}{M} \frac{\partial^2 \psi}{\partial X \partial r} + \frac{m_e}{M} \frac{\partial^2 \psi}{\partial r \partial X} + \frac{\partial^2 \psi}{\partial r^2} \\ &\Rightarrow \frac{\partial^2}{\partial x_e^2} = \frac{m_e^2}{M^2} \frac{\partial^2}{\partial X^2} + \frac{2m_e}{M} \frac{\partial^2}{\partial X \partial r} + \frac{\partial^2}{\partial r^2} \end{aligned} \quad (4)$$

$$\begin{aligned} \frac{\partial \psi}{\partial x_N} &= \left(\frac{\partial \psi}{\partial X} \right) \left(\frac{\partial X}{\partial x_N} \right) + \left(\frac{\partial \psi}{\partial r} \right) \left(\frac{\partial r}{\partial x_N} \right) = \frac{\partial \psi}{\partial X} \cdot \frac{m_N}{M} + \frac{\partial \psi}{\partial r} \cdot (-1) \\ &= \frac{m_N}{M} \frac{\partial \psi}{\partial X} - \frac{\partial \psi}{\partial r} \end{aligned}$$

$$\begin{aligned} \frac{\partial^2 \psi}{\partial x_N^2} &= \frac{\partial}{\partial x_N} \frac{\partial \psi}{\partial x_N} = \frac{m_N}{M} \cdot \frac{\partial}{\partial X} \left(\frac{m_N}{M} \frac{\partial \psi}{\partial X} - \frac{\partial \psi}{\partial r} \right) - \frac{\partial}{\partial r} \left(\frac{m_N}{M} \frac{\partial \psi}{\partial X} - \frac{\partial \psi}{\partial r} \right) \\ &= \frac{m_N^2}{M^2} \frac{\partial^2 \psi}{\partial X^2} - \frac{m_N}{M} \frac{\partial^2 \psi}{\partial X \partial r} - \frac{m_N}{M} \frac{\partial^2 \psi}{\partial r \partial X} + \frac{\partial^2 \psi}{\partial r^2} \\ &\Rightarrow \frac{\partial^2}{\partial x_N^2} = \frac{m_N^2}{M^2} \frac{\partial^2}{\partial X^2} - \frac{2m_N}{M} \frac{\partial^2}{\partial X \partial r} + \frac{\partial^2}{\partial r^2} \end{aligned} \quad (5)$$

b)

$$\begin{aligned}
\hat{H} &= -\frac{\hbar^2}{2m_e} \frac{\partial^2}{\partial x_e^2} - \frac{\hbar^2}{2m_N} \frac{\partial^2}{\partial x_N^2} - \frac{e^2}{4\pi\epsilon_0 \underbrace{|x_e - x_N|}_{=|r|}} \\
&= -\frac{\hbar^2}{2m_e} \left(\frac{m_e^2}{M^2} \frac{\partial^2}{\partial X^2} + \frac{2m_e}{M} \frac{\partial^2}{\partial X \partial r} + \frac{\partial^2}{\partial r^2} \right) - \frac{\hbar^2}{2m_N} \left(\frac{m_N^2}{M^2} \frac{\partial^2}{\partial X^2} - \frac{2m_N}{M} \frac{\partial^2}{\partial X \partial r} + \frac{\partial^2}{\partial r^2} \right) \\
&\quad - \frac{e^2}{4\pi\epsilon_0 |r|} \\
&= -\frac{\hbar^2 m_e}{2M^2} \frac{\partial^2}{\partial X^2} - \frac{\hbar^2}{M} \frac{\partial^2}{\partial X \partial r} - \frac{\hbar^2}{2m_e} \frac{\partial^2}{\partial r^2} - \frac{\hbar^2 m_N}{2M^2} \frac{\partial^2}{\partial X^2} + \frac{\hbar^2}{M} \frac{\partial^2}{\partial X \partial r} - \frac{\hbar^2}{2m_N} \frac{\partial^2}{\partial r^2} \\
&\quad - \frac{e^2}{4\pi\epsilon_0 |r|} \\
&= -\frac{\hbar^2}{2M^2} \underbrace{(m_e + m_N)}_{=M} \frac{\partial^2}{\partial X^2} - \frac{\hbar^2}{2} \underbrace{\left(\frac{1}{m_e} + \frac{1}{m_N} \right)}_{=\frac{1}{\mu}} \frac{\partial^2}{\partial r^2} - \frac{e^2}{4\pi\epsilon_0 |r|} \\
&= -\frac{\hbar^2}{2M} \frac{\partial^2}{\partial X^2} - \frac{\hbar^2}{2\mu} \frac{\partial^2}{\partial r^2} - \frac{e^2}{4\pi\epsilon_0 |r|}
\end{aligned} \tag{6}$$

c) Setzen den Produktansatz $\Psi(X, r) = \xi(X)\psi(r)$ in die Schrödingergleichung(6)

$$\hat{H}\xi(X)\psi(r) = E\xi(X)\psi(r)$$

$$-\frac{\hbar^2}{2M} \frac{\partial^2}{\partial X^2} \xi(X)\psi(r) - \frac{\hbar^2}{2\mu} \frac{\partial^2}{\partial r^2} \xi(X)\psi(r) - \frac{e^2}{4\pi\epsilon_0 |r|} \xi(X)\psi(r) = E\xi(X)\psi(r) \tag{7}$$

$$-\frac{\hbar^2}{2M} \psi(r) \frac{d^2}{dX^2} \xi(X) - \frac{\hbar^2}{2\mu} \xi(X) \frac{d^2}{dr^2} \psi(r) - \xi(X) \frac{e^2}{4\pi\epsilon_0 |r|} \psi(r) = E\xi(X)\psi(r) \tag{8}$$

Beide Seiten durch $\xi(X)\psi(r)$ teilen

$$\begin{aligned}
\frac{1}{\xi(X)} \cdot \left(-\frac{\hbar^2}{2M} \frac{d^2}{dX^2} \xi(X) \right) + \frac{1}{\psi(r)} \cdot \left(-\frac{\hbar^2}{2\mu} \frac{d^2}{dr^2} \psi(r) \right) + \frac{1}{\psi(r)} \left(-\frac{e^2}{4\pi\epsilon_0 |r|} \psi(r) \right) &= E \\
\underbrace{\frac{1}{\xi(X)} \left(-\frac{\hbar^2}{2M} \frac{d^2}{dX^2} \xi(X) \right)}_{=E_X} + \underbrace{\frac{1}{\psi(r)} \left[\left(-\frac{\hbar^2}{2\mu} \frac{d^2}{dr^2} - \frac{e^2}{4\pi\epsilon_0 |r|} \right) \psi(r) \right]}_{=E_r} &= E \\
\implies -\frac{\hbar^2}{2M} \frac{d^2}{dX^2} \xi(X) &= E_X \xi(X)
\end{aligned} \tag{9}$$

und

$$\left(-\frac{\hbar^2}{2\mu} \frac{d^2}{dr^2} - \frac{e^2}{4\pi\epsilon_0 |r|} \right) \psi(r) = E_r \psi(r) \tag{10}$$

d) $M \approx m_N$ und $\mu \approx m_e$ weil $m_N \gg m_e$.

2. Das Teilchen im Zentralpotential

$$\left(\left(\frac{\hat{p}_r^2}{2\mu} + \frac{\hat{l}^2}{2\mu r^2} \right) + V(r) \right) \Psi(r, \theta, \phi) = E \Psi(r, \theta, \phi) \quad (11)$$

$$\hat{p}_r = -i\hbar \left(\frac{\partial}{\partial r} + \frac{1}{r} \right)$$

a) Einsetzen den Separationsansatz $\Psi_{n,l,m_l}(r, \theta, \phi) = R_{n,l}(r)Y_{l,m_l}(\theta, \phi)$ in die Schrödinger-gleichung (11)

$$\left[\frac{\hat{p}_r^2}{2\mu} + \frac{\hat{l}^2}{2\mu r^2} + V(r) \right] R_{n,l}(r)Y_{l,m_l}(\theta, \phi) = E R_{n,l}(r)Y_{l,m_l}(\theta, \phi) \quad (12)$$

$$Y_{l,m_l}(\theta, \phi) \frac{\hat{p}_r^2}{2\mu} R_{n,l}(r) + R_{n,l}(r) \frac{\hat{l}^2}{2\mu r^2} Y_{l,m_l}(\theta, \phi) + Y_{l,m_l}(\theta, \phi) V(r) R_{n,l}(r) = E R_{n,l}(r) Y_{l,m_l}(\theta, \phi) \quad (13)$$

Lösung des Winkelanteils bekannt: $\hat{l}^2 Y_{l,m_l}(\theta, \phi) = \hbar^2 l(l+1) Y_{l,m_l}(\theta, \phi)$

$$Y_{l,m_l}(\theta, \phi) \frac{\hat{p}_r^2}{2\mu} R_{n,l}(r) + R_{n,l}(r) \frac{\hbar^2 l(l+1)}{2\mu r^2} Y_{l,m_l}(\theta, \phi) + Y_{l,m_l}(\theta, \phi) V(r) R_{n,l}(r) = E R_{n,l}(r) Y_{l,m_l}(\theta, \phi) \quad (14)$$

Multiplizieren beide Seiten der Gleichung mit $Y_{l,m_l}^*(\theta, \phi)$ und integrieren zwischen den Grenzen

$$\therefore \int_0^\pi \int_0^{2\pi} Y_{l,m_l}^*(\theta, \phi) Y_{l,m_l}(\theta, \phi) d\theta d\phi = 1$$

Die Gleichung für den Radianteil $R_{n,l}(r)$ ergibt

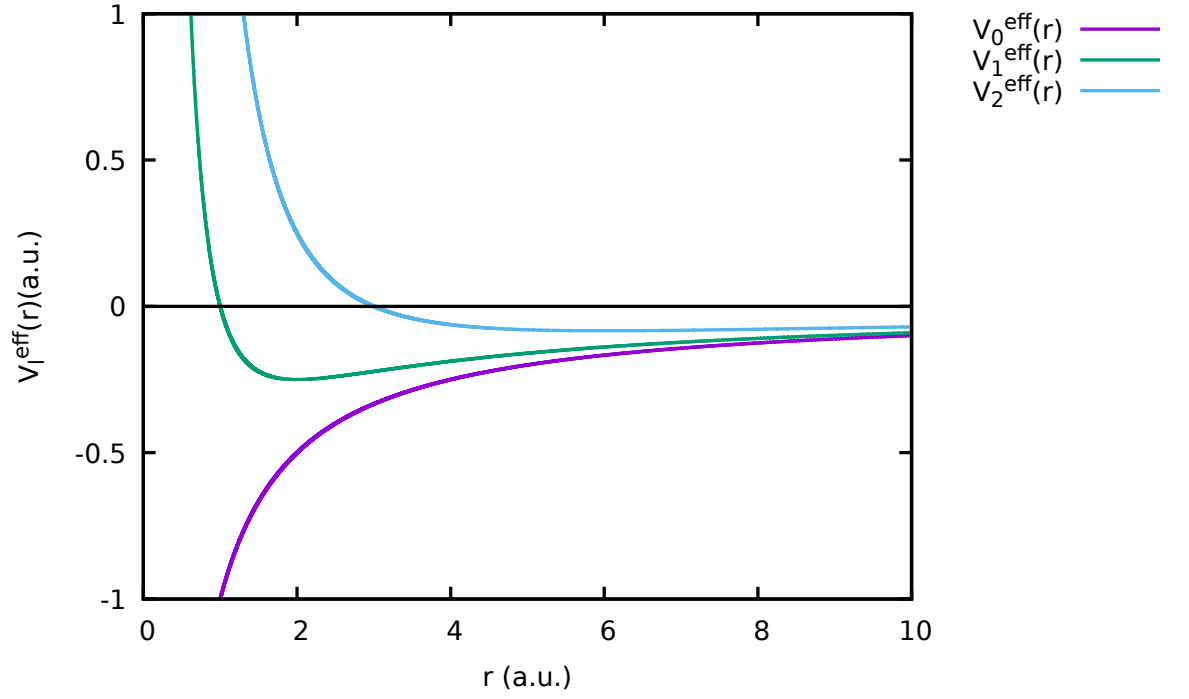
$$\frac{\hat{p}_r^2}{2\mu} R_{n,l}(r) + \frac{\hbar^2}{2\mu r^2} l(l+1) R_{n,l}(r) + V(r) R_{n,l}(r) = E R_{n,l}(r) \quad (15)$$

$$\therefore \left(\frac{\hat{p}_r^2}{2\mu} + \frac{l(l+1)\hbar^2}{2\mu r^2} + V(r) \right) R_{n,l}(r) = E R_{n,l}(r) \quad (16)$$

b)

$$V_l^{eff}(r) = V(r) + \frac{\hbar^2 l(l+1)}{2\mu r^2} = -\frac{e^2}{4\pi\epsilon_0 r} + \frac{\hbar^2 l(l+1)}{2\mu r^2}$$

In atomaren Einheiten, $e = 1$, $\frac{1}{4\pi\epsilon_0} = 1$, $\hbar = 1$, $\mu \approx 1$



c)

$$\Psi_{n,l,m_l}(r, \theta, \phi) = R_{n,l}(r)Y_{l,m_l}(\theta, \phi) \implies |\Psi_{n,l,m_l}(r, \theta, \phi)|^2 = |R_{n,l}(r)|^2|Y_{l,m_l}(\theta, \phi)|^2$$

$$R_{n=1,l=0}(r) = \frac{2}{a_0^{\frac{3}{2}}} e^{-\frac{r}{a_0}} \implies |R_{n=1,l=0}(r)|^2 = \frac{4}{a_0^3} e^{-\frac{2r}{a_0}} \quad (17)$$

$$Y_{l=0,m_l=0}(\theta, \phi) = \frac{1}{2\sqrt{\pi}} \implies |Y_{l=0,m_l=0}(\theta, \phi)|^2 = \frac{1}{4\pi} \quad (18)$$

(Referenz: Tabelle 3.2, Kapitel 3, *Molecular Quantum Mechanics*, 5th ed., Peter

Atkins und Ronald Friedman)

$$\begin{aligned}
P &= \int_0^{r'} \int_0^\pi \int_0^{2\pi} r^2 \sin \theta dr d\theta d\phi |\Psi_{n,l,m_l}(r, \theta, \phi)|^2 \\
&= \int_0^{r'} \int_0^\pi \int_0^{2\pi} r^2 \sin \theta dr d\theta d\phi |R_{n=1,l=0}(r)|^2 |Y_{l=0,m_l=0}(\theta, \phi)|^2 \\
&= \frac{1}{4\pi} \int_0^{r'} \frac{4}{a_0^3} e^{-\frac{2r}{a_0}} r^2 dr \int_0^\pi \sin \theta d\theta \int_0^{2\pi} d\phi \\
&= \frac{1}{\pi a_0^3} \int_0^{r'} r^2 e^{-\frac{2r}{a_0}} dr \cdot [-\cos \theta]_0^\pi \cdot [\phi]_0^{2\pi} \\
&= \frac{1}{\pi a_0^3} \cdot 2 \cdot 2\pi \int_0^{r'} r^2 e^{-\frac{2r}{a_0}} dr = \frac{4}{a_0^3} \int_0^{r'} r^2 e^{-\frac{2r}{a_0}} dr \\
&= \frac{4}{a_0^3} \left[\frac{r^2 e^{-\frac{2r}{a_0}}}{-\frac{2}{a_0}} \right]_0^{r'} - \frac{4}{a_0^3} \int_0^{r'} \frac{2r e^{-\frac{2r}{a_0}}}{-\frac{2}{a_0}} dr \\
&= -\frac{a_0}{2} \cdot \frac{4}{a_0^3} \left(r'^2 e^{-\frac{2r'}{a_0}} - 0 \right) + \frac{4}{a_0^2} \int_0^{r'} r e^{-\frac{2r}{a_0}} dr \\
&= -\frac{2}{a_0^2} r'^2 e^{-\frac{2r'}{a_0}} + \frac{4}{a_0^2} \left[\frac{r e^{-\frac{2r}{a_0}}}{-\frac{2}{a_0}} - \frac{e^{-\frac{2r}{a_0}}}{\frac{4}{a_0^2}} \right]_0^{r'} \\
&= -\frac{2}{a_0^2} r'^2 e^{-\frac{2r'}{a_0}} + \frac{4}{a_0^2} \left[\frac{r' e^{-\frac{2r'}{a_0}}}{-\frac{2}{a_0}} - \frac{e^{-\frac{2r'}{a_0}}}{\frac{4}{a_0^2}} - 0 + \frac{e^0}{\frac{4}{a_0^2}} \right] \\
&= -\frac{2}{a_0^2} r'^2 e^{-\frac{2r'}{a_0}} + \frac{4}{a_0^2} \left[-\frac{a_0 r' e^{-\frac{2r'}{a_0}}}{2} - \frac{a_0^2 e^{-\frac{2r'}{a_0}}}{4} + \frac{a_0^2}{4} \right]
\end{aligned} \tag{19}$$

Für den Fall $r' = 2a_0$

$$\begin{aligned}
P &= -\frac{2}{a_0^2} \cdot 4a_0^2 e^{-4} + \frac{4}{a_0^2} \left[\frac{2a_0^2 e^{-4}}{2} - \frac{a_0^2 e^{-4}}{4} + \frac{a_0^2}{4} \right] \\
&= 1 - 13e^{-4} = 0.76
\end{aligned}$$

Dies ist das 1s-Orbital des Wasserstoffatoms. Die Aufenthaltswahrscheinlichkeit P ist unabhängig von den Winkelvariablen (θ, ϕ) , da die sphärischen Harmonischen für das 1s-Orbital konstant sind.