Molecular Dynamics Simulations

Searching Configuration Space

Energy minimization (EM)



They are always interested in the minimum !!!

Searching Configuration Space

Energy minimization (EM) Molecular dynamics (MD)





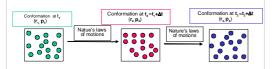
They are always interested in the minimum !!!

Molecular Dynamics: As a technique to search configuration space

The kinetic energy present in the systems allows it to surmount the energy barrier that are in the order of $k_{\rm B}T$ per degree of freedom



The Molecular Dynamics method generates configurations applying Nature's Law of motions for atoms of a molecular system.



Solve Newton's equations of motion =

Time evolution of the system

Nature's law of motion

Newton's Equations of Motion

Force = Mass x Acceleration

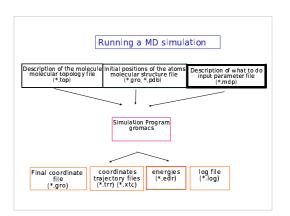
or Acceleration = Force/Mass



$$F_i = -\frac{\partial}{\partial r_i} V(r_1, r_2, \dots, r_N)$$

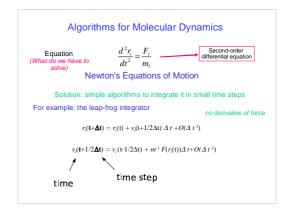
What governs the outcome of a MD simulation?

- •starting configuration
- •choice of degrees of freedom
- •force field parameters
- •boundary conditions
 •treatment of non-bonded interactions
- •integration time step
- •treatment of temperature and



Keywords for non-bonded interactions and periodic boundary conditions in *.mdp file

| Interactions | Coulombtype = cut-off | treatment of electrostatic interactions | coulombtype = cut-off | ;Periodic Boundary Conditions pbc = xyz ;Periodic Boundary Conditions



Choosing integration time step

What is an appropriate time step for the integration?







hannens			
	System	Types of motion	Suggested time step
	Rigid molecules	translation, rotation	5 fs
	Flexible molecules, rigid bonds	translation, rotation, torsions	2 fs
	Flexible molecules, flexible bonds	Translation, rotation, torsions, vibrations	1 or 0.5 fs

Keywords for MD algorithms in *.mdp file

title = MD simulation of ;Perform an MD simulation for 20 ps (example)

 $integrator = md \qquad ; Algorithm \ (md = molecular \ dynamics)$

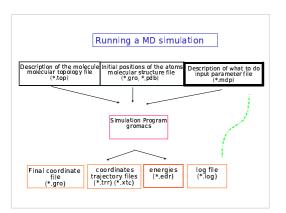
= 0.0 ; starting time in picoseconds

= 0.002 ;time step in picoseconds dt

nsteps = 10000 ;number of steps (total simulation time)

;constraints

constraints = all-bonds ;Bond types to replace by constraints



Keywords for output file control in *.mdp file

;Parameters to save data (what to write and when); frequency

nstxout nstvout = 100 ;coordinates to output trajectory file (*.trr) = 100 ;velocities to output trajectory file(*.trr)

nstlog = 10 ;energies to log file(*.log) nstenergy = 10 ;energies to energy file (*.edr)

nstxtcout = 10 ;coordinates to xtc trajectory (*.xtc)

;group(s) to write

xtc_grps = BUT SOL ;to xtc trajectory (*.xtc)

energygrps = BUT SOL ; to energy file (*.edr)

Temperature and Pressure

Standard MD: number of particles N, volume V and total energy $E_{\rm tot}$ of the system are conserved (NVE ensemble)

$$E_{tot} = E_{potential} + E_{kinetic}$$

- Experiments are often performed at constant pressure P and temperature T (NPT ensemble)
 Approximations (cut-offs, integration) can lead to instabilities and thus energy and/or temperature variations

Temperature and pressure control needed

Temperature

* Temperature T related to the kinetic energy K of the system and therefore to the velocities

$$K = \sum_{i=1}^{N} \frac{1}{2} \vec{n} \vec{v}_i^2 = \sum_{i=1}^{N} \frac{p_i^2}{2m_i} = \frac{k_B T}{2} (3N - N_c)$$

- =3N-N $_{c}$ is the number of degrees of freedom in the
- In particular N is the number of particles and N, is the number of constraints on the system
- The temperature can thus be controlled by modifying the velocities of atoms and molecules in the system

Berendsen thermostat

- Temperature control by weak coupling to an external reference temperature bath
 - Velocities are scaled at a rate proportional to the temperature difference

$$dT \, I \, dt = \tau_T^{-1} \left(T_{\, bath} - T \left(t \right) \right) \qquad \text{\mathbf{t} is a coupling parameter}$$

- Exponential decay toward reference temperature
- The scaling factor λ becomes

$$\lambda = (1 + \Delta t / \tau_T (T_{bath} / T(t) - 1))^{-1/2}$$

At integration time step

Pressure

- The pressure is related to the volume of the system and the interaction between particles
- Ideal gas: PV = Nk_bT
- Real system: $PV = Nk_bT \frac{1}{3k_bT} \sum_{i=1}^{N} \sum_{j=1}^{N} f_{ij}$ The second term is the virial describing the contribution due to the forces between the particles
- Can easily be calculated in MD when calculating the forces
- Pressure can be controlled by varying the volume of the system and scaling the position of the molecules

Berendsen barostat

Pressure can be controlled in a similar way as T by weak coupling to an external pressure bath with the rate of change given by:

$$dP/dt = \tau_{p}^{-1}(P_{-d} - P(t))$$

Pressure can adjusted by scaling the volume by a factor λ or the coordinate by $\lambda^{\rm Li3}$

 $\lambda = 1 - \kappa \Delta t / \tau_P (P_{ref} - P(t))$

where K is the iso-thermal compressibility

Keywords for temperature and pressure control in *.mdp file

| Temperature coupling at T=333 K (example) | Temperature coupling scheme | Temperature coupling scheme | Temperature coupling scheme | Temperature coupling scheme | Temperature | Tem

| Pressure coupling Prouple | Service | Pressure coupling scheme | Service | Pressure relaxation time | Compressibility | 4.5e-5 | Compressibility | 4.5e-6 | Pressure relaxation time | Compressibility | Compres

Example of a simulation setup

System setup

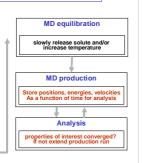
Starting structure

Initial Energy Minimization in chosen force field in vacuum Generate solvent box

Energy Minimization of solvent, solute fixed

Add counter ions if needed

Energy Minimization of solvent + ions, solute fixed



MD simulations of butane in water solution at constant temperature and pressure

•Create the parameter file for around 500 ps simulation (using 2 fs time step) at the following simulation conditions:
a) Constant temperature (T=300 K);
Constant pressure (p= 1 atm);
b) Constant temperature (T=350 K);
Constant pressure (p= 1 atm).

•Run one MD simulation for each temperature.

Tips:
Use different directories for each simulation. Use top to check that your simulations are running Use df to check that disk space

Files we need for the MD simulation in water of butane at constant temperature and pressure.

•Geometry file (*.gro or *.pdb): the output file of the minimizations performed in water solution: confout.gro.

•Corrected topology file for all-atom force field (butane top)
PS: include "water.itp" and check the total number of water molecule at the end of the file
•Water topology file, water.itp.
•Force field files (ff.itp and ffib.itp)

•Parameter file for molecular dynamics simulation in water
Fill the file md.mdp with the correct keywords

Check that the molecular dynamics simulation is equilibrated.

*Use the tools g energy to extract the **temperature** and the **pressure** of the system from the energy file (ener.edr) using "g energy -f ener.edr - o t.xvg" and "g energy -f ener.edr - o p.xvg" Plot t.xvg and p.xvg file (output of g energy) using "xmgrace t.xvg" and "xmgrace p.xvg".

•After how many ps do the **temperature** and the **pressure** reach a plateau? How large is the temperature at t=0 ps ?

*Use the tools g_energy to extract total energy from the energy file "g_energy-fenered" o-energy_xvg".
*Plot energy xvg file (output of g_energy) using "xmgrace energy xvg". After how many ps does the potential energy reach a plateau".
*Is your simulation equilibrated before \$00 ps? If not, run longer, if yes, after how many picoseconds?

•Visualise the trajctory using the software vmd (traj.xtc)